

Common Bonding Patterns

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Element	Hybridization			Bonds	Unshared e ⁻ Pairs	Valence e ⁻
	sp ³	sp ²	sp			
Carbon, C		$\left[\begin{array}{c} + \\ \\ \diagup \\ \\ \diagdown \\ C \end{array} \right]$ carbocation		3	0	6
	$\begin{array}{c} \\ -C- \\ \end{array}$	$\begin{array}{c} \\ C= \\ \end{array}$	$\begin{array}{c} \equiv C- \\ =C= \end{array}$	4	0	8
	$\begin{array}{c} \ominus \\ \\ -C- \\ \\ \ddot{\cdot} \end{array}$ alkyl carbanion	$\begin{array}{c} \ominus \\ \\ \ddot{\cdot} C= \\ \end{array}$ vinyl anion	$\begin{array}{c} \equiv C:\ominus \\ \quad \\ \quad \ddot{\cdot} \end{array}$ acetylidy anion	3	1	8
Nitrogen, N						
	$\begin{array}{c} \\ -N^+ \\ \end{array}$ ammonium cation	$\begin{array}{c} \\ N^+= \\ \end{array}$	$\begin{array}{c} \begin{array}{c} + \\ \\ \equiv N- \end{array} \\ \begin{array}{c} \\ + \\ =N= \end{array} \end{array}$	4	0	8
	$\begin{array}{c} \ddot{\cdot} \\ \\ -N- \\ \end{array}$	$\begin{array}{c} \ddot{\cdot} \\ \\ N= \\ \end{array}$	$\begin{array}{c} \equiv N: \\ \quad \\ \quad \ddot{\cdot} \end{array}$	3	1	8
Oxygen, O						
		$\left[\begin{array}{c} \ddot{\cdot} \\ =O^+ \\ \\ H \end{array} \right]$ protonated carbonyl	$\left[\begin{array}{c} + \\ \\ \equiv O: \end{array} \right]$ oxonium cation	3	1	8
	$\begin{array}{c} \ddot{\cdot} \\ \\ -O: \end{array}$	$\begin{array}{c} \ddot{\cdot} \\ \\ O: \end{array}$		2	2	8
Halogen, X (F, Cl, Br, I)						
		$\left[\begin{array}{c} \ddot{\cdot} \\ -X^+ \\ \end{array} \right]$ halonium cation		2	2	8
	$\begin{array}{c} \ddot{\cdot} \\ \\ -X: \end{array}$			1	3	8
		$\begin{array}{c} \ddot{\cdot} \\ \\ :X^- \end{array}$ halide anion		0	4	8